### **Interreg** North Sea Region NuReDrain



European Regional Development Fund





### Webinar, 24 November 2022



- Please mute yourself.
- Feel free to ask questions in the chat.
- The webinar will be recorded.
- Handouts will be put available afterwards.



- Nutrient Removal and Recovery from Drainage water
- **1**/3/2017 30/9/2021 -> 30/6/2023
- Interreg North Sea Region
- Project cost: € 3.516.378 Fund: € 1.758.187
- II partners in 3 countries -> 7 partners in 3 countries (BE, DE, DK)



## **Eutrophication status**



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- Red : problem areas
- Orange : potential problem areas
- Green : non-problem areas

Source: OSPAR

## **Nutrient sources**





Source: European Environment Agency

## **Nutrient sources**





Source: European Environment Agency



### I. Source based measures

- Reduce nutrient input (optimize fertilization)
- Reduce losses from soil (adapt rotations, grow catch crops, manage crop residues, ...)



### End-of-pipe filter systems



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## 2 water types



# I. Drainage water



**Bigger volumes** 

Low N and P concentrations

Rain dependent

Periodic

## 2. Greenhouse effluent



Smaller volumes

High N and P concentrations

Controllable

Continuous or periodic

### **Field cases**



**Field sites for filter tests Phosphorus removal** 0 Nitrogen removal • Fensholt Phosphorus & Nitrogen Taastrup removal . Warnow River Basin Lethe River Venner Bruchkanal Middelkerke Zedelgem Destelbergen De Blankaart Nevele Heusden Onze-Lieve-Vrouw-Waver Staden Roeselare Anzegem Belgie / Belgique Main





I. Long term evaluation and upscaling of iron coated sand filters for P removal from agricultural drainage water

Junwei Hu (Ghent University)

- 2. Can filter materials be regenerated and reused? Jef Bergmans (VITO)
- 3. Combining filter systems for N and P removal from greenhouse wastewater

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# Long term evaluation and upscaling of iron coated sand filters for P removal from agricultural drainage water

Junwei Hu, Hui Xu, Stany Vandermoere, Stefaan De Neve Department of Environment Ghent University







# Why is P removal important?





reduced fertilizer (only to be possible on the long run)

## What do we need?



There are no efficient management practices to reduce P leaching on the short term.



17—40 % is drained in NW Europe a large mitigation potential

installing P filters at the end of the drains could be potential solution

- Reduce P loads as much as possible (< 0.1 ppm, Water Framework Directive)
- Process discontinuous and peak flows
- Low cost and easy to be installed

## **Phosphorus Filter Development**



### Phosphorus Sorbing Materials: Iron coated sand (ICS)

Lab scale P sorption capacity and speed: ICS > bauxite > glauconite > olivine = biotite.



### By-product from drinking-water industry

Vandermoere S., Ralaizafisoloarivony N., Van Ranst E., De Neve S. (2018). Reducing phosphorus (P) losses from drained agricultural fields with iron coated sand (- glauconite) filters. Water Research, 141, 329–339.

# Simple bucket filter and prototype

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### Simple bucket filter



#### Prototype







# Filter performance



### P removal efficiency

- Volume weighted average TP concentration inlet
- Volume weighted average DRP concentration inlet
- Volume weighted average TP concentration outlet
- Volume weighted average DRP concentration outlet



TP: total phosphorus DRP: dissolved reactive phosphorus



	Price [€]	Life span [years]
Filter bucket	600	15
ICS materials	50	2
Labor for installation	50 (self-installation) /100 (external-installation)	2
Total [€/year]	90-115	



- + Low-tech solution: easy installation and operation
- + High P removal efficiency
- + Low cost of filter materials
- + Causes no other contaminations
- + No impact on accessability and landscape

- + Only applicable for individual drain
- + Mostly remove dissolved reactive P

### **Recent Progress**





# Long term performance of filter box for single drain



### Dig-in filter box for single drain



Upscaled filter box for larger volume of water



4 existing P filter boxes
Anzegem, 2018-11-29
Zedelgem, 2018-11-30
Zedelgem 1mm (2), 2020-1-10
Zedelgem 2mm (3), 2020-1-10

## Long term performance



P removal efficiency Zedelgem Anzegem ■ TP-in ■ TP-out ■ DRP-in ■ DRP-out ■ TP-in ■ TP-out ■ DRP-in ■ DRP-out 0,50 0,50 36% P concentration (mg P/L) 43% 38% 0,40 0,40 20% 19% 30% 23% 0,30 0,30 4% 60% 54% 0,20 0,20 87% 43% 99% 73% 56% 57% 0,10 0,10 0,00 0,00 2018-2019 2019-2020 2020-2021 2021-2022 2018-2019 2019-2020 2020-2021 2021-2022 0.04-3,6 0.1-7,2 0.1-2,5 0.07-1,9 0.4-10,5 0.4-6,3 0,4-17,6 0.9-12,3 m³/d m³/d m³/d m<sup>3</sup>/d m<sup>3</sup>/d m<sup>3</sup>/d m³/d m<sup>3</sup>/d



P removal efficiency



## **Dig-in filterbox**



The P filterbox is currently placed in the ditch but ideally the ditch is not to be obstructed.









## **Dig-in filter box**



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Staden dig-in (1)

Staden dig-in (2)

### The installation was proven to be feasible

## **Dig-in filterbox**



### P removal efficiency



Given the P concentration is low, the dig-in P filter boxes are functioning



To treat bigger volume of drainage water

### Collector drain



### Open pond



## **Upscaled P filterbox**



- Capable of treating bigger volume of drainage water
- 140 L iron coated sand





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### Godelieve fields of Inagro

### Middelkerke (Belgian coast)







### P removal efficiency



Flow: 7.2-16.9 m3/d

Flow: 17.0-36.2 m3/d

## The next



- Effectiveness vs flow
- Clogging of water from open pond
- Electricity supply















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# Thanks!









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# **Desorption of ICS grains**

Jef Bergmans – VITO



The opportunity to regenerate filter material enables that high-cost materials can be reused several times decreasing the total cost of operation.

- protocol for the regeneration P saturated filter material
- scale-up of the process
# Characterisation of saturated ICS grains





### Mostly Fe, Si and P Also C, Mg and Ca found



# Literature



Different strategies are available in literature

- Acidic route  $\rightarrow$  focus on P recovery
- Alkaline route → focus on **filter material recovery**

The alkaline route regenerates the hydroxyl groups

 $\text{Fe-OH} + \text{H}_2(\text{PO}_4)^- \rightarrow \text{Fe-O-H}_2(\text{PO}_3) + \text{OH}^-$ 

A significant amount of OH- ions are needed to reverse the equilibrium and the excess OH<sup>-</sup> ions are slowly release afterwards changing the pH of the eluate

Several studies available in literature providing valuable information, However:

- Static small scale desorption experiments
- Test performed on artificially loaded sorbents





0.1 M – contact time 60 s

Re-use of the stripping solution also gives similar results for lower base concentration Improved P desorption by changing contact time needs further investigation

# Scale-up







Optimisation at small scale (10 g)

Larger scale (1000 g)

# Scale-up



Recirculation with 0.1 M NaOH Liquid/Solid (L/S) ratio = 10 Flow 60 ml/min





5.2 mg P desorbed per g ICS Equilibrium after 18 recirculations < 1 ppm Fe in the solution

# Scale-up



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- Keeping pH at pH 12.5
- 25 liter for 12 kg

Larger scale (12 kg)









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- Desorption of ±6.0 g P /kg sorbent
- Desorbed sorbents sent to PCS after washing step (water) and pH decrease



## NuReDrain: results 2022





//IIIN PCS

© PCS | Meeting 4-5 October 2022 NuReDrain Roeselare, Belgium

# Sorption at pH = 12.5



- L/S = 5
- Desorption of 10.7 g P /kg sorbent
- Higher pH: damaging ICS grain integrity







Precipitation of phosphate With 1 g Ca(OH)<sub>2</sub> per liter

 $\Rightarrow$  Precipitation product over filtration: 0.96 g

First analysis results:

- Mainly calcium phosphate carbonates [Ca<sub>5</sub>(PO<sub>4</sub>)<sub>3</sub>(CO<sub>3</sub>)<sub>0.5</sub>]: ±75%
- About 25% CaCO<sub>3</sub>/Ca(OH)<sub>2</sub>
- Trace levels of Fe, Si, Zn





- High pH solution: ±12.5 or lower (mixing with washing and/or acidic solution)
  - Phosphate: 25 g/l
  - Calcium: 0.7 g/l
  - Iron: 0.3 mg/l
  - Trace levels of Si, Zn
  - Sodium?
  - Desorption is also possible with KOH



- Desorption of P is possible for the saturated ICS
  - Type of base has neglible influence (KOH vs NaOH)
  - Recirculation of solution is possible
  - Automated addition of extra alkaline solution
- Higher alkalinity gives better P removal but
  - A balance will have to be found between ICS integrity and P recovered capacity



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## NuReDrain 2.0 Webinar:

# Combining filter systems for N and P removal from greenhouse wastewater

### Els Pauwels (PCS Ornamental Plant Research)

#### N-removal: constructed wetlands + MBBR





### MBBR: Biological denitrification on bio-carriers



- Carbon-source
- Aeration pump



Carbon-source





### NuReDrain: results 2022



#### • PCS: constructed wetlands





NuReDrain: results 2022

• PCS: MBBR





■ storage pond ■ MBBR

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### **P-Removal P-filters**





VITO B

ICS\_UGent

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- Testing capacity saturated ICS grains from UGent - Testing regenerated ICS

grains from VITO

### **Filter-material P-filters**



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ICS



PCS: P-Filters







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Waterflow in 2022

In 2020-2021: MBBR + P-filter - adaptations were needed

In 2022: first P-filter and then N-filter ICS were replaced





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Drainwater from greenhouse with azalea

- Buffer tank 5 m<sup>3</sup>
- P filter
- MBBR
- Daily volume: 1,5 m<sup>3</sup>









In buffer tank
After filters (1/8m) EON 100 8/08/22 5/09/22 4/10/22 31/10/22

### **Microflor - Orchid**









#### 1. What is a MBBR?

A Moving Bed Biofilm Reactor (or MBBR for short) removes nitrogen from water by converting nitrate into nitrogen gas by means of biological processes. A MBBR consists of a tank filled with water, in which plastic carriers are located that are set in motion (Photo IPhoto 1). The irregular and large specific surface area of the carriers forms an ideal habitat for various micro-organisms (Photo 2Photo 2). On these carriers grows active sludge (biofilm) and this carriers out the denification.

A MBBR requires little maintenance and is simple to construct yourself with the help of this information sheet.



Photo 1: Set-up of Moving Bed Biofilm Reactor (MBBR) at PCS Ornamental Plant Research

Information sheet: "How do I build my own MBDRY" - Version date 2005/2020 Dears up it concerns with the interme for date 2006/2014 (Charlenaux). No and of the publication may be reproduced without the prior within permittion of PCS.

# Manual available: Dutch and English

- ✓ Simple and compact
- ✓ Processes 3 m³/day
- ✓ Robust
- ✓ Cheap
- ✓ No recovery
- ✓ Add C-source (5-8%)
- ✓ Difficult operation in winter





# **Questions?**

Thanks to: Marijke Dierickx Bruno Jacobson Da Silva

E: <u>Els.Pauwels@pcsierteelt.be</u> T: +32 9 353 94 88



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# Cost effectiveness of nutrient removal filters at a catchment scale

Andreas Bauwe, Bernd Lennartz – University of Rostock

NUREDRAIN 2.0 WEBINAR November 24 2022



# Filter types



a) NuReDrain applications

Moving Bed Bioreactor (MBBR) N



# b) other applications

Woodchip filter N + P



### Phosphorus filter P



### Constructed Wetland N + P



The main study questions were:

- 1. What is the **nitrate removal potential** using filters at catchment scale and
- 2. what are the **annual costs**?



Necessary data

- Mean annual waterflow of the drainage pipe Q (m<sup>3</sup>/yr)
- Mean annual NO<sub>3</sub><sup>-</sup>-N concentration in the drainage water C (mg/L)
- Removal efficiency of the filter (%)
- Costs (€/ kg N reduction)

Calculations

$$\begin{aligned} F_{actual}\left(\frac{kg}{yr}\right) &= Q\left(\frac{m^3}{yr}\right) * c\left(\frac{kg}{m^3}\right) \\ F_{filter}\left(\frac{kg}{yr}\right) &= F_{actual}\left(\frac{kg}{yr}\right) * \left(1 - \frac{Removal \ efficiency \ (\%)}{100 \ (\%)}\right) \\ Costs\left(\frac{\epsilon}{yr}\right) &= \left(F_{actual}\left(\frac{kg}{yr}\right) - F_{filter}\left(\frac{kg}{yr}\right)\right) * Costs\left(\frac{\epsilon}{kg}\right) \end{aligned}$$
# Example 1: Drainage plot





### **Drain plot characteristics**

Size: 4.2 ha Q = 39 m³/d or 7056 m³/yr c = 9.6 mg NO<sub>3</sub><sup>-</sup>-N/L

### Assumptions

Removal efficiency: 30 – 90%

Costs: MBBR:  $50 - 500 \notin$ kg reduction WF:  $2 - 20 \notin$ kg reduction SFW:  $5 - 95 \notin$ kg reduction



# **Example 1: Drainage plot**



Nomograms to calculate annual costs based on varying removal efficiencies and installation/maintenance costs



Total Costs: 339 €/year

Costs: 45 €/kg N red. Total Costs: 1524 €/year Costs: 250 €/kg N red. Total Costs: 8467 €/year





### Assumptions

Annual discharge: 80 - 150 mm/yr

NO<sub>3</sub><sup>-</sup>-N concentrations: 3 - 15 mg/L

Implementation of filter technologies: NO<sub>3</sub>-N concentrations >10 mg/L

- MBBR: small-sized drainage plots (<2 ha) at costs of 50 500 €/kg N reduction
- Woodchip filters: medium-sized drainage plots (2-10 ha) at costs of 2 20 €/kg N reduction
- Surface-flow constructed wetlands: large-sized drainage plots (>10 ha) at costs of 5 - 95 €/kg N reduction

Removal efficiencies: 30 - 90%

### Calculations

All values will be randomly selected within the given ranges

100 runs were conducted

# **Example 2: Catchment**









- Due to many hardly predictable costs and necessary simplifications, single calculations are mostly to be understood as rough estimates.
- Therefore, scenario analyses using **multiple calculations** can provide a realistic range of expected costs and nutrient reductions that policymakers and stakeholders can work with at the watershed level.
- The results suggest that filter systems, if installed widely, can help to reduce nutrient pollution in surface waters significantly.
- The installation and maintenance of filter systems can be very costly.
- Filter systems will always be side-measures and will not replace the necessity of sustainable agricultural systems.













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# Nutrient filter systems in practice: construction manuals, fact sheets and cost assessment

Charlotte Boeckaert, Vlakwa

# **Nuredrain information**



field descriptions sheets case videos materials filter fact articles



NuReDrain, Interreg VB North Sea Region Programme

# **Project videos**

### North Sea Region NuReDrain

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#### Phosphorus filter box for drainage water in agricultural fields

Tests on filter systems for phosphorus removal from drainage water in agricultural fields by <u>Ghent University</u> and <u>Inagro</u>



### Phosphorus filtering with a sediment and reactive filter in Denmark

Video about the phosphor removal filter system of <u>Aarhus University</u> and <u>Copenhagen</u> <u>University</u> in <u>Fensholt</u>



#### Modelling the reduction potential of filters at catchment scale

Video about <u>Rostock University</u>'s model that estimates the reduction potential of nutrient removal filters at catchment scale



#### Nitrogen and phosphorus removal in horticulture

Filter systems for Nitrogen and Phosphorus removal in horticulture by <u>PCS</u> and <u>KU</u> Leuven



#### Phosphorus filter system for a water production centre

Video about tests on a phosphorus filter system in a water production centre by <u>De</u> <u>Watergroep</u> and <u>VITO</u>



#### Mobile constructed wetland for nitrate removal Video about the field test of a denitrication filter system by QOWV at Lether river in

Lower Saxony



#### A zero valent iron filter for drainage water

Copenhagen University is testing an experimental Zero Valent Iron filter for the removal of phosphorus and nitrogen from drainage water.



Inline phosphorus filter system for drainage water Video about the field test of a phosphorus filter system by <u>LWK Niedersachsen</u> for drainage water at Venner Bruchkanal in Lower Saxony



#### Relive the NuReDrain webinars

NuReDrain Webinars

results.

In September and October 2020 NuReDrain organised a series of webinars about the research

Webinar I : phosphorus filter systems for drainage water



### **Fact sheets**





Ntrate-enriched drainage water is diverted through a bioreactor filled with woodchips, which provides anosic and carbon-enriched conditions. Microorganisms in the bioreactor break down nitrate to nitrogen gas, which is relaxed in the atmosphere. As not rate represents the major form of nitrogen in drainage water, nitrogen loads to the aquatic ecosystems are significantly reduced.





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**NuReDrain** 

Zero-valent iron and zeolite filter Phosphate and nitrate removal and nitrogen recycling	Price: c 3860 Plan: 1, 5 m/d PQ, removal N03 removal Plant Protection Product removal OM removal
Benefits 🛛 🔒	Limitations 🧃
Nitrate can be completely removed, even at lo concentrations and low temp. Ammonium can be recovered enabling nitroge to be recycled and the recovered enabling nitroge to be recycled of the removed Iron(0) formed during 2VI corrosion contribute to futher increase the reactivity of the system	passivating ZVI corrosion layers Oxygen in drainage water will also consume ZV Reduction of water consumes ZVI and generates H <sub>i</sub>
Ammonium is capturee in another mee Phosphate removal: When nitrate is re oxide) is a super filter for phosphate	nt is gane on the first value around. the mark received in the second s
Zero-valent iron and	ct details: Adrian Florea (aff@plen.ku.dk) Hans Christian (haha@plen.ku.dk) Price < 3800 Florea < and
Zero-Valent Iron and zeolite filter Nitrate removal by biological denitrification	PO4, removal NO3 removal Plant Protection Product removal OM removal



# Filter construction manual



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2. Dimensioning



### 3. DIY building



Water	Filter	CAPEX	ΟΡΕΧ	Yearly cost	Total P removal (kg P)	Cost effectiveness (€/kg P)
Drainage (0,25 mg P/I)	P filterbox	€ 635	€ 8,62	€ 702	0,06	1.304
	Drainage w	ater (0,46 m	g P/I)		0,19	Ļ
	Drainage w	ater (0,12 m	g P/I)		0,02	Î
Greenhouse (15 mg P/I)	DIY	€ 690	€ 9,10	€ 291	1,94	327

	Application	CAPEX	ΟΡΕΧ	Yearly cost	Total N removal (kg P)	Cost effectiveness (€/kg N)
DIY	Greenhouse effluent	€ 3.000	€ 1.131	€ 1.400	12,44	113
Subsoil	Drainage	€ 30.000	€ 3.260	€ 6.000	52,84	114
Containerized	Drainage Off-grid	€ 50.000	€ 1.867	€ 6.034	44,44	136
	Drainage	€ 40.900	€ 2.967	€ 6.315	44,44	142





- I. Source based measures
- 2. Identification of nutrient hot spots
- 3. Set reduction goals
- 4. Calculate costs
- 5. Prioritize locations
- 6. Install filter systems

# **Acknowledgements**



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# Thank you for your attention!

